Molecular Dynamics Refinement of a Thermitase–Eglin-c Complex at 1.98 Å Resolution and Comparison of Two Crystal Forms that Differ in Calcium Content

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The crystal structure of the complex of thermitase with eglin-c in crystal form II, obtained in the presence of 5 mm-CaCl₂, has been determined at 1.98 Å resolution. The structure was solved by a molecular replacement method, then molecular dynamics crystallographic refinement was started using the thermitase–eglin-c structure as determined for crystal form I. A ten degrees rigid body misplacement of the core of eglin-c was corrected by the molecular dynamics crystallographic refinement without any need for manual rebuilding on a graphics system. A final crystallographic R-factor of 16.5% was obtained for crystal form II.

The comparison of the complexes of thermitase with eglin-c in the two crystal forms shows that the eglin-c cores are differently oriented with respect to the protease. The inhibiting loop of eglin binds in a similar way to thermitase as to subtilisin Carlsberg. A tryptophanyl residue at the S_4 site explains the preference of thermitase for aromatic residues of the substrate at the P_4 site. The difference in the P_1 binding pocket, asparagine in thermitase instead of glycine in subtilisin Carlsberg, does not change the binding of eglin-c. The preference for an arginyl residue at the P_1 site of thermitase can be explained by the hydrogen bonding with Asn170 in thermitase.

Three ion-binding sites of thermitase have been identified. The strong and weak calciumbinding sites resemble the equivalent sites of subtilisin Carlsberg and subtilisin BPN', though there are important amino acid differences at the calcium-binding sites. The medium-strength calcium-binding site of thermitase is observed in the subtilisin family for the first time. The calcium is bound to residues from the loop 57 to 66. A difference in chelation is observed at this site between the two crystal forms of thermitase, which differ in calcium concentration. Additional electron density is observed near Asp60 in crystal form II, which has more calcium bound than form I. This density is possibly due to a water molecule ligating the calcium ion or the result of Asp60 assuming two significantly different conformations.

1. Introduction

(a) Thermitase

Thermitase is an extracellular thermostable serine protease, isolated from *Thermoactinomyces vulgaris* (Frömmel & Höhne, 1981). The primary sequence of this protein, which is 279 amino acid residues long,

has been determined by Meloun et al. (1985). Sequence comparisons clearly indicate that thermitase is a member of the subtilisin family of serine proteases, with the Asp-His-Ser catalytic triad. X-ray structures have been determined of subtilisin BPN' at high (McPhalen et al., 1985a; Katz & Kossiakoff, 1986; Bott et al., 1988) and at moderate

resolution (Wright et al., 1969; Drenth et al., 1972). of subtilisin Carlsberg (Bode et al., 1987; McPhalen et al., 1985b; Neidhart & Petsko, 1988), as well as of mutants of these proteins (Bryan et al., 1986; Katz & Kossiakoff, 1986). Also, a high-resolution structure is available for proteinase K (Betzel et al., 1988a). Significant differences exist within this family in, for example, enzyme catalysis (Brömme et al., 1986; Kleine, 1982) and thermostability (Voordouw et al., 1976; Frömmel & Höhne, 1981). Thermitase is known to be an extremely thermostable member of the subtilisin family (Frömmel & Höhne, 1981) and has been the subject of a number of investigations in our laboratories (Dauter et al., 1988; Kelders et al., 1989; Gros et al., 1989). A preliminary structure of the native enzyme has been presented by Teplyakov et al. (1989). The crystallographic refinement of thermitase complexed with eglin-c in a different crystal form, called crystal form I, has been reported by Gros et al. (1989). The initial solution of the structure of crystal form II has been described by Dauter et al. (1988).

The degree of sequence similarity between several subtilisins is given in Table 1. Thermitase is included in a subtilisin subgroup that contains a single free cysteinyl residue. This free cysteine is located near the active site, as can be seen from the sequences, and may play a role in enzyme catalysis. Binding a Hg-derivative to this cysteine inactivates the protein (Teplyakov et al., 1986). A large number of amino acid differences and a smaller number of insertions and deletions are apparent comparing the thermitase sequence with that of subtilisin Carlsberg. The largest difference between thermitase and the other subtilisins occurs at the N terminus. The N terminus of thermitase is extended by seven residues. A totally new arrangement of the N terminus in thermitase is observed in comparison to the other subtilisins, as revealed by the previous crystallographic studies on thermitase (Gros et al., 1989): glutamine 2 of subtilisins Carlsberg and BPN', which is a ligand of a strongly bound calcium ion, is replaced by Asp5 from the N-terminal extension in thermitase. Proteinase K, in contrast, lacks the N-terminal calcium-binding site (Betzel et al., 1988b). This illustrates the divergence in evolution at this site of the subtilisins.

The termostability of subtilisins is strongly

Table 1
Sequence identities (%) between members of the subtilisin family and their cysteine content

	Sub. Carlsberg	Sub. BPN'	Thermitase	$\begin{array}{c} \text{Proteinase} \\ \text{K} \end{array}$
Sub. Carlsberg	100	69	47	37
Sub. BPN'	69	100	47	37
Thermitase	47	47	100	36
Proteinase K	37	37	36	100
No. of Cys	0	0	1	5
No. of disulphide bridges	0	0	0	2

influenced by the presence of calcium ions (Voordouw et al., 1976). This phenomenon is observed also for thermitase (Frömmel & Höhne, 1981). However, unlike subtilisin Carlsberg and BPN', thermitase binds three calcium ions (Frömmel & Höhne, 1981) instead of two. The most weakly bound calcium ion stabilizes the enzyme against thermal denaturation or autoproteolysis (Frömmel & Höhne, 1981). These authors report that the other two calcium ions bind much more strongly and protect against autolysis and thermal inactivation in extreme conditions. Pantoliano et al. (1988) have reported three crystal structures of subtilisin BPN', which differed in CaCl₂ content of the crystallization medium. Although the calcium concentration strongly influences the rate of thermal inactivation of this subtilisin, only minor main-chain or sidechain alterations were found. For proteinase K, however, Bajorath et al. (1989) report a concerted set of small movements due to binding of calcium in the strong calcium-binding site. This site is common to all subtilisin structures determined so far, and is known as the weak calcium-binding site in subtilisin Carlsberg and BPN'. The function of the calciumbinding sites, the local changes due to amino acid differences amongst the different serine proteases, and the location and structure of the "new" calcium-binding site in thermitase are of considerable interest.

Crystal form II of the complex of thermitase with eglin-c, discussed in this paper, has been grown in the presence of 5 mm-calcium chloride (Dauter et al., 1988). Crystal form I of the same complex (Gros et al., 1989), was obtained from a medium free of calcium chloride. The thermitase–eglin-c models of crystal form I and II will be referred to as TE1 and TE2, respectively. Comparison of the three calciumbinding sites of thermitase in the two different crystal forms should give insight into the role of these binding sites with respect to thermostability and protection against autoproteolysis.

(b) Eglin-c

Eglin-c is a 70 amino acid residue protein inhibitor, isolated from the leech Hirudo medicinalis (Seemüller et al., 1981). This small protein is a very potent inhibitor of subtilisin, chymotrypsin, and the human proteinases elastase and cathepsin G (Seemüller et al., 1986). These inhibiting characteristics make eglin-c a potential pharmaceutical tool. Although eglin-c lacks cysteinyl residues, it obeys the mechanism of inhibitory action against serine proteases (Seemüller et al., 1986) described by Laskowski & Kato (1980). Stabilization of the scissile bond occurs via Arg51 and Arg53 in this cysteine-independent inhibitor, as observed in crystal structures of eglin-c with subtilisin Carlsberg (Bode et al., 1986; McPhalen et al., 1985a). None of the crystal structures of eglin-c complexed with a serine protease reported so far (Bode et al., 1986; McPhalen et al., 1985a; Gros et al., 1989) shows a reduced electron density at the reactive site peptide bond. This indicates that proteolytic cleavage of the inhibitor is not a prerequisite for, or consequence of, inhibitory action. As in the previous studies on subtilisins (Bode et al., 1986; McPhalen et al., 1985a) and thermitase (Gros et al., 1989), eglin-c was used to prevent autodegradation of the serine protease during crystallization. By studying eglin-c in the various complexes, detailed insight is gained about its interesting inhibitory characteristics.

(c) Crystallographic molecular dynamics refinement

The application of the molecular dynamics algorithm (Hermans, 1985; Karplus & McCammon, 1981) in biomacromolecular crystallographic refinement was first described by Brünger et al. (1987). Fujinaga et al. (1989) have implemented the crystallographic refinement in the simulation package GROMOS (van Gunsteren & Berendsen, 1987) analogous to Brünger et al. (1987). The reduction of the required model building on a computer graphics system in the refinement process is substantial when using this new method (Brünger, 1988; Fujinaga et al., 1989; Gros et al., 1989). Two types of strategy have been reported that are suitable for obtaining large "automatic" model corrections by molecular dynamics (MD†) refinement (Brünger, 1988; Fujinaga et al., 1989; Gros et al., 1989). Brünger (1988) proposes a simulated annealing (SA) protocol for using the crystallographic molecular dynamics refinement. In contrast, we performed MD refinement as in our previous studies on the thermitase-eglin-c complex, form I (Gros et al., 1989); that is, during the refinement the upper resolution limit of the X-ray diffraction data was gradually increased, concomitant with increasing X-ray weights, while working at relatively low temperatures (600 and 300 K; Gros et al. (1989), in comparison to temperatures over 1000 K used by Brünger (1988)). No additional positional restraint of the molecules in the crystal structure model is necessary, nor any adjustment of the stereochemical force fields, when using this low-temperature protocol (Fujinaga et al., 1989; Gros et al., 1989). Of major interest in the present work is the discovery that the MD refinement procedure was able to perform a substantial rigid body motion fully automatically.

2. Materials and Methods

(a) Crystal growth

Thermitase was prepared as described (Frömmel et al., 1978) and was provided for this study by Dr W. Höhne of the Humboldt University, Berlin. Eglin-c was a kind gift

from Dr H. P. Schnebli of Ciba-Geigy, Basel.

The thermitase–eglin-c crystals were obtained as described (Dauter et~al., 1988). Briefly, crystals were grown by hanging-drop vapour diffusion in Linbro tissue-culture dishes. The hanging drops contained 5 mg thermitase/ml, 1.7 mg eglin-c/ml, 37.5 mm-sodium acetate, 5 mm-CaCl₂ and 5% (w/v) PEG 4000 at pH 5.2, 20°C. The drops were $15~\mu$ l in volume. The molar ratio of enzyme to inhibitor was 1:1.3. A total of 60 drops were set up. In most dishes there were no crystals, but in each of 2 drops, a single large crystal with elongated tetrahedral morphology, about 0.8 mm in each dimension, grew on the surface of the glass.

(b) Data collection and reduction

One of the 2 large crystals was used for characterization of the space group on the precession camera (Dauter *et al.*, 1988). The space group, as for crystal form I, was $P2_12_12_1$, but the cell dimensions were quite different, $a=49\cdot3$ Å, $b=67\cdot3$ Å and $c=90\cdot5$ Å, with $V_{\rm m}$ of $2\cdot1$ Å 3 /dalton (1 Å $=0\cdot1$ nm).

The 2nd crystal of crystal form II was used to collect 3-dimensional data to 1·8 Å using the EMBL beam-line X31 at the DORIS storage ring, DESY, Hamburg. The data were recorded at room temperature, on CEA-25 reflex X-ray film with 3 films/pack using a modified Arndt–Wonacott oscillation camera. A total of 90° of data were measured, the crystal rotating around the c-axis. Exposure times/pack were 10 to 15 min. Many reflections at low resolution were saturated on the films in the 1·8 Å data set. The same crystal was therefore used to record 90° of low-resolution (2·8 Å) data. Exposure times were 1 to 2 min. Owing to lack of crystals, no data were recorded for the blind region.

Films were digitized using an Optronics Photoscan P-1000 microdensitometer in the absorbance range 0 to 3 and a 50 μ m raster size. The MOSFLM film processing package was used in evaluating the films. The 1·8 Å data presented considerable problems during integration, owing to the crystal slipping during data collection. This was explained by the rather generous amount of mother liquor left in the capillary to ensure that the single available crystal of form II did not dry out. Eight different orientation matrices were finally required for successive groups of films. The data processing is summarized in Table 2, and the 1·8 Å data had a final merging R-factor $(R_{\rm sym} = \Sigma \Sigma |I - I_i|/\Sigma \Sigma I_i)$, where I is the mean value of i intensity measurements I_i) of 8·2 %. All data, including

Table 2
Summary of the data collection for crystal form II

Parameter	High- resolution set	Low- resolution set	High + Low
Wavelength (Å)	1.15	1.488	
Crystal-to-film (mm)	56.0	90.4	
Number of films	72×3	40×3	112
Oscillation range (°)	1.25	2.25	
Resolution limit (Å)	1.8	2.8	1.8
Refins measured	84,568	27.266	94,298†
Refins unique	24,789	7611	26,664
$R_{\text{sym}} \left(\frac{\%}{0} \right) \ddagger$	8.2	4.2	7.2

[†] This number is less than the sum of the low and high-resolution data, as the data from the high-resolution set were excluded for resolution less than 3.5 Å.

[†] Abbreviations used: MD, molecular dynamics; SA, simulated annealing; EMX, energy minimization, including X-ray restraints; MDXREF, molecular dynamics, including X-ray restraints; BFAC, individual temperature factor refinement; CE, subtilisin Carlsberg—eglin-c complex; TE1, thermitase—eglin-c complex form I; TE2, thermitase—eglin-e complex form II.

 $[\]label{eq:resolvent_sym} \ddagger R_{\mathrm{sym}} = \Sigma \Sigma |I - I_i| / \Sigma \Sigma I_i.$

negative intensity observations, were included in the calculation of the R-factor.

Before collection of the 2·8 Å data, the crystal was dried more carefully in the capillary. As a result, no further slippage was detected and a single orientation matrix was used for the whole 90° of data. The $R_{\rm sym}$ for the 2·8 Å data was 4·2%. The final merging of all the data for the 2 limiting resolutions is summarized in Table 2, and gave an R-factor of 7·2% for 26,664 reflections making up 93% of the complete data to 1·8 Å; 88·4% of these had an intensity greater than 3 standard deviations. Data from the high-resolution set were excluded from the final merging if they had a resolution less than 3·5 Å.

The intensities were converted to structure factor amplitudes using the programme TRUNCATE (French & Wilson, 1978). Inspection of the $R_{\rm sym}$ and the fraction of observed to unobserved reflections led to the data to 1.98 Å only being used in the refinement described in this paper.

(c) Molecular replacement and first refinement cycles

The phase problem was solved by molecular replacement using the fast rotation function and R-factor translation search programmes from the CCP4 package. We first used the unrefined co-ordinates of subtilisin Novo (Drenth et al., 1972) from the Brookhaven data bank (Bernstein et al., 1977) as a model. With these, we were unable to find a satisfactory solution.

We subsequently used refined co-ordinates of the subtilisin Carlsberg-eglin-c complex kindly provided by Professor M. James, Edmonton (McPhalen et al., 1985b). The orientation and position of the complex in the unit cell were determined unambiguously from the 1st calculation with this model, in spite of the relatively low level of homology (47% identity) between the 2 enzymes. For both the rotation and translation searches, data in the resolution range 3.0 to 10.0 Å were used, with Patterson integration range limits of 3 and 25 Å. About 1200 reflections were used in the rotation function. The highest peak, which corresponded to the correct orientation of the molecule in the cell, was 114 on an arbitrary scale. The next highest peak in the map was 75; 931 reflections were used for the translation function. The lowest, and correct, minimum had an R-factor of 37.3%, the next lowest was 42% and the average value in the map was 47.6%. The relevant sections of the rotation and translation functions are shown in Fig. 1 of Dauter et al. (1988). These results show the great advantage of using accurate, highly refined co-ordinates in molecular computations

The model of subtilisin Carlsberg-eglin-c was placed in the cell of crystal form II of thermitase-eglin-c. It was refined using Hendrickson-Konnert restrained leastsquares minimization (Hendrickson & Konnert, 1981), with the use of fast Fourier routines to calculate electron densities and gradients (Finzel, 1987). In 20 cycles using data from 5 to 3 Å, the R-factor dropped from 51.5 to 30.2%. FRODO (Jones, 1978) was then used, on an Evans & Sutherland PS330 graphics system, to place the majority of the correct side-chains for thermitase in the model. A further 16 cycles of least-squares minimization with data from 5 to 2.8 Å resolution gave an R-factor of 25.0%, and an additional 30 cycles with 5 to 2.5 Å data gave a final R-factor of $26\cdot2\%$. The eglin had much less well-defined density than the thermitase at this stage. The reason for this is discussed below. At this point, the restrained refinement of crystal form II was stopped. The refined model of crystal form I, TE1 (Gros et al., 1989),

was superimposed on the current model of crystal form II, TE2, using C^{α} co-ordinates only, and the refinement continued with the combined use of X-ray data and molecular dynamics.

(d) Molecular dynamics refinement procedures

The crystallographic refinement was performed using the molecular simulation package GROMOS (van Gunsteren & Berendsen, 1987). This suite of programs has been modified by Fujinaga et al. (1989) to include crystallographic refinement. Both energy minimization (EMX), including X-ray restraints, and molecular dynamics (MDXREF), also including X-ray restraints, have been used in the refinement. For individual temperature factor refinement (BFAC), the program $FFTR\bar{E}F$ of Agarwal (1978) has been used. Typically, the EMX and MDXREF runs consisted of 100 to $\bar{2}50$ steps. One step in MDXREF corresponded to a 2fs time-step. During MDXREF, the system was coupled to a constant heat bath (Berendsen etal., 1984). The interaction parameters of GROMOS for covalent bond, angle, dihedral, improper dihedral, Lennard-Jones and electrostatic interactions for simulation in vacuo were used without any alteration (van Gunsteren & Berendsen, 1987). The cut-off radius for calculating electrostatic interactions was set to 8 $\mbox{\normalfont\AA}$. The weight for the X-ray energy was approximately equal to the observed root-mean-square difference in $|F_{\rm obs}|$ and $|F_{\rm calc}|$ during the refinement and is depicted in Fig. 1(b). The limit for maximal co-ordinate shift of any atom before recalculating $F_{\rm calc}$ values was dependent on the higher resolution limit of the X-ray data used in the refinement. At 3.0 Å resolution, this was set to 0.2 Å and at 1.98 Å to 0.1 Å. Between recalculations, the harmonic approximation, as given by Jack & Levitt (1978), was used to update the X-ray gradient. The calculations for EMX, MDXREF and BFAC were performed on a CONVEX C1-XP of the BIOSON Research Institute.

An overview of the weights and the resolution range used for EMX, MDXREF and BFAC is given in Fig. 1(b). First, the starting model of the complex was subjected to $300~{\rm steps}$ of EMX, using diffraction data between $8{\cdot}0$ and 3.0~Å resolution. The individual atomic temperature factors from TE1 were used as starting temperature factors for TE2. Throughout the refinement, all reflections with $|F_{\text{obs}}| > \sigma_{|F_{\text{obs}}|}$ within a certain resolution shell were used. The weight on the crystallographic energy term was gradually increased during these 300 steps. After these initial 300 steps of EMX, 400 steps of MDXREF were performed at 600 K; first at 3.0 Å resolution and after 200 steps at 2.5 Å resolution (see point A in Fig. 1). Before the temperature factor refinement, the structure was refined at 2.5 Å resolution towards a local minimum by EMX. All individual B factors were reset to 12.0 Å^2 before the temperature factor refinement. The resolution was increased in the following EMX runs to 2.2 Å and later to 1.98 Å, which is the upper resolution limit of the current data set. After co-ordinate refinement at 1.98 Å resolution, the individual temperature factors were refined; this time, the full resolution range from 8.0 to 1.98 Å was used. Up to this point in the refinement, point C in Fig. 1, no manual rebuilding was carried out.

Here, an electron density map at 1.98 Šresolution was calculated. Using this map, the model was refitted and water positions were identified by a procedure to be described below. The temperature factor was reset to 12.0 Ų for those parts of the model that were refitted. The refinement was continued using 8.0 to 1.98 Å resolution data for all subsequent steps. After 100 steps of

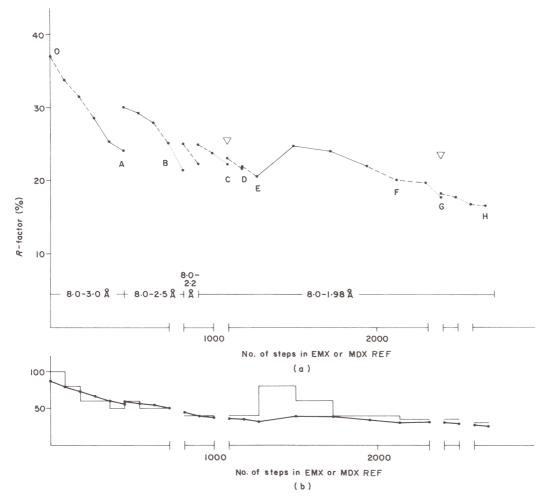
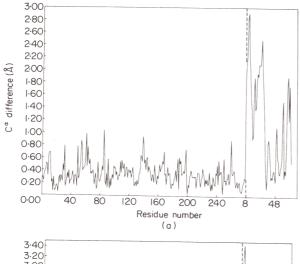


Figure 1. (a) The course of the crystallographic R-factor during the refinement. The R-factor $(R = \Sigma ||F_{\rm obs}|-|F_{\rm calc}||/\Sigma ||F_{\rm obs}||$) as a function of the number of steps in the refinement is given. Molecular dynamics refinement (MDXREF) is indicated by continuous lines, energy minimization with X-ray restraints (EMX) is given by broken lines and the individual temperature factor refinement is indicated by dots. One step in MDXREF corresponds to a 2fs time-step. Labels indicate points along the refinement referred to in the text. The triangles indicate model building sessions. The resolution range used is shown at the bottom. (b) The root-mean-square difference between $|F_{\rm obs}|$ and $|F_{\rm calc}|$ is given in thick lines. $\sigma_{\rm x}$, used for the calculation of the weight of the X-ray term $w_{\rm x}$, is given in thin lines. $w_{\rm x} = 1/\sigma_{\rm x^2}$ and $|F_{\rm calc}| = w_{\rm x} \Sigma (|F_{\rm obs}|-|F_{\rm calc}|)^2$ (Brünger et~al., 1987; Fujinaga et~al., 1989).

EMX, point D in Fig. 1, the located water molecules were added to the model, all with $B = 14.0 \text{ Å}^2$. Another 100 steps of EMX followed. Here, we chose to reduce the X-ray restraints and use MDXREF again. Remaining potential model errors have a higher chance of being corrected if the restraints are relaxed (Gros et al., 1989). Therefore, MDXREF was performed with a lower weight on the X-ray energy. Gradually, the weight was increased to its previous value. The temperature in MDXREF was this time kept at 300 K, because of the presence of weakly bound water molecules. Before the 2nd and last model building step, the temperature factors were further refined. The last manual corrections were applied to the model and an additional 44 water molecules were positioned; see point G in Fig. 1. This time, the newly found water molecules were given a B of 16.0 Å^2 . The resulting model was energy minimized, by 100 steps of EMX. Subsequently, the final individual temperature factors were determined and the co-ordinates further refined by 100 steps of EMX. The ions were identified on the basis of distances to ligands, number of hydrogen bond acceptors as ligands and, for 2 of the 3 sites, homology to other

subtilisins. After identification of the ions present in the crystal structure, a few additional steps of EMX were performed with the ions included in the model.

Altogether, 2 sessions of model refitting were carried out on an Evans & Sutherland Picture System 390, using the interactive display program FRODO (Jones, 1978). The density maps used were all calculated with coefficients $(2m|F_{obs}|-D|F_{calc}|)$, $\exp(i\alpha_{calc})$. As shown by Read (1986), the model bias due to co-ordinate errors is suppressed when these coefficients are used. The stereochemical geometry of the model was analysed by calculating the energy/atom throughout the molecules using a program kindly provided by Dr J. Postma. Only the residues that contained energy terms more than 3 to 4σ above the mean energy were inspected and refitted according to the electron density. The water molecules were positioned by locating peaks in the difference density (with electron coefficients map $\langle c_{alc}| \rangle$, exp(i α_{calc})), using the program PEKPIK $(m|F_{obs}|-|F$ from the XTAL package (Hall & Stewart, 1987). Only positions with distances between 2.0 and 3.5 Å from any other atom in the model were accepted.



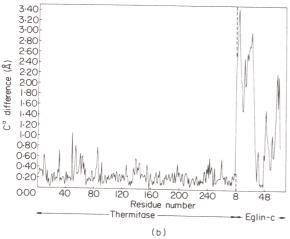


Figure 2. (a) The difference (Å) in C^{α} position between beginning and end of the molecular dynamics refinement (point 0 and point H in Fig. 1) as function of residue number. (b) The difference in C^{α} position between the refined models of TE1 and TE2 as function of residue number. The residue numbering of eglin-c starts with 8, because the 1st 7 residues are invisible in the map and are omitted from the model.

3. Results

(a) $Molecular\ dynamics\ refinement:\ accommodation$ of rigid body motion

The general decrease of the crystallographic R-factor during the refinement is depicted in Figure 1(a). After superposition of TE1 on the preliminary model of the complex in crystal form II, the R-factor was 37.0% for data between 8.0 and 3.0 Å resolution; see point 0 in Figure 1(a). A gradual drop in R-factor was observed during the EMX, MDXREF and BFAC runs. Before the first model building session, point C in Figure 1, the R-factor had dropped to $22\cdot2\%$ for data between $8\cdot0$ and 1.98 Å resolution. At point C, the electron density of the model was inspected on a computer graphics system. From the energy analysis, areas with bad geometries were identified. This mainly involved side-chains, such as valine and isoleucine. The worst geometries in the model at point C

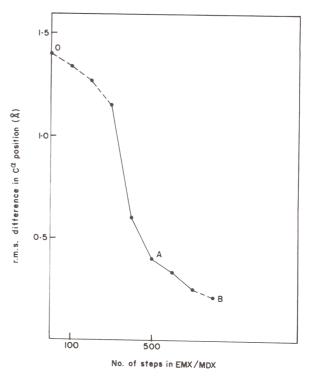


Figure 3. The root-mean-square (r.m.s.) difference in C^{α} positions of eglin-c between intermediate models and the final model shown as a function of the number of steps in EMX or MDXREF needed to obtain the intermediate model. Continuous lines indicate root-mean-square differences during MDXREF runs, and broken lines refer to EMX runs. The labels 0, A and B correspond with the labels in Fig. 1.

occurred at Asp62 and Asp188. Asp62 is involved in Ca²⁺ binding, as will be discussed later. Asp188 was known to be less well modelled in the TE1 model, because of its deviating ϕ, φ combination (Gros et $\it al., 1989$). In these parts, the model was corrected to fit the electron density as well as possible. The structure was further refined, including 183 water molecules from point D onwards. After point E, the X-ray restraints were relaxed, and thereafter gradually reinforced again to the old value. This process resulted in a decrease in R-factor of 0.5% with slightly lower total energies for the stereochemical restraints. At point G in Figure 1, an R-factor of 18.0% was obtained for data between 8.0 and 1.98 Å resolution. At this point, 44 new water positions were identified, and the electron density of the model was inspected. The refitting at point G involved: (1) the side-chain of Thr120, from staggered to eclipsed conformation; (2) the sidechains of Val200, Val234, Val335, all from eclipsed to staggered conformation; (3) a slight readjustment of the side-chain of Argl1; and (4) a total refitting of Lys280. By EMX, BFAC and EMX, a final R-factor at point H in Figure 1 of $16.5\,\%$ was obtained, for 19,730 reflections between 8.0 and 1.98 Å resolution, with $|F_{\text{obs}}| > \sigma |F_{\text{obs}}|$.

Most interestingly, a major model correction had occurred during the refinement, although the

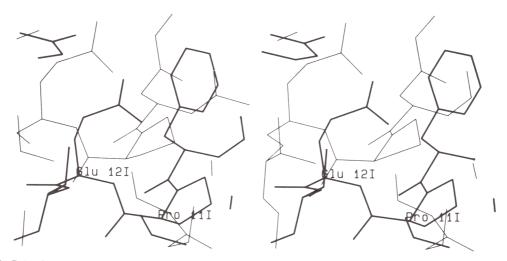


Figure 4. Initial position in thin lines and final position in thick lines of Pro11I and Glu12I of eglin-c giving an impression of the large shifts carried out automatically by the molecular dynamics refinement procedure.

smooth improvement of the R-factor, as shown in Figure 1, does not indicate anything striking. Figure 2(a) shows the difference in C^{α} positions for thermitase and eglin-c between point 0 and point H in Figure 1(a). The final differences in C^{α} positions after superposition of the thermitases of the fully refined models is shown in Figure 2(b). The difference in the C^{α} positions of eglin-c is remarkably large. This error in the starting model, at point 0, appeared to correspond to a considerable rigid body motion of the eglin-c core with respect to thermitase.

Figure 3 shows the root-mean-square difference in C^{α} positions for the 63 observed C^{α} atoms of the eglin-c model during the first steps in the refinement with respect to the final C^{α} positions. The rootmean-square deviation between the C^{α} positions of initial and final model was 1.40 Å. The first three runs of EMX brought this down to only 1.15 Å. The effect of increasing the weight (see Fig. 1) on the X-ray energy is reflected in the slopes of these three runs. We refrained from continuing EMX with even stronger weights on the X-ray restraints, because of the increase in the energies for the stereochemical terms during the last 100 steps. Instead, MDXREF was started using the same resolution range, data from 8.0 to 3.0 Å resolution, and the same X-ray weight, $\sigma_x = 60$. The root-mean-square error in C^{α} positions of eglin-c dropped from 1·15 Å to 0·60 Å within the first 100 steps of MDXREF. The second 100 steps of MDXREF with $\sigma_x = 50$ reduced the root-mean-square error to 0.40 Å. Refinement at 2.5 Å resolution brought the error within the limits of accuracy corresponding to the resolution limit; that is, 0.21~Å after MDXREF and EMX with 2.5~Åresolution as upper limit for the data.

The region of eglin-c, where the deviations are among the largest between the starting model (point 0 in Fig. 1) and the final model (point H in Fig. 1), is shown in Figure 4. Clearly, the errors in the initial model are severe. For example, the starting position of ProllI is closer to the correct position of Glu12I

than to its own correct position. The initial position of the C^{α} of Prol1I was 2·2 Å away from the final position of the C^{α} of Glu12I, whereas its own final position was 2·8 Å away from the initial position. For many atoms of eglin-c that are far away from the inhibiting loop, shifts up to 3 Å were performed automatically by the molecular dynamics procedure.

(b) Accuracy and analysis of the model

A good indicator of the stereochemical correctness of the main-chain folding of a protein is provided by a Ramachandran plot. Figure 5 displays the angles for thermitase and eglin-c. The allowed regions for non-glycine residues are indicated according to Ramakrishnan & Ramachandran (1965). Large clusters are found at the right-handed α -helix and the β -strand regions. The largest deviation from ideality in φ, ϕ combination for a non-glycine residue is found for Asp38. This is the active site aspartate, which is part of the catalytic triad. Its uncommon φ,ϕ angles are also observed in the other subtilisin structures (Bode et al., 1987; McPhalen et al., 1985a,b; Betzel et al., 1988b), and relate to its catalytic function. There are in total four residues (slightly) outside the strictly allowed regions commonly observed in thermitase and subtilisin Carlsberg; Gln19, Asp38, Asp185 and Thr217. These deviating residues, except Asp38, are all located in short turns. This is probably the reason for the minor deviation in torsion angles from the allowed regions. These deviating φ, ϕ angles of thermitase have been obtained after a substantial exploration of conformational space, which gives the deviations additional relevance. Overall, the Ramachandran plot does not indicate any serious error in the final model of the complex of thermitase and eglin-c.

For the final model of the complex, the energies, as well as the mean energies of the various energy terms, used in the molecular dynamics refinement are given in Table 3. The root-mean-square devia-

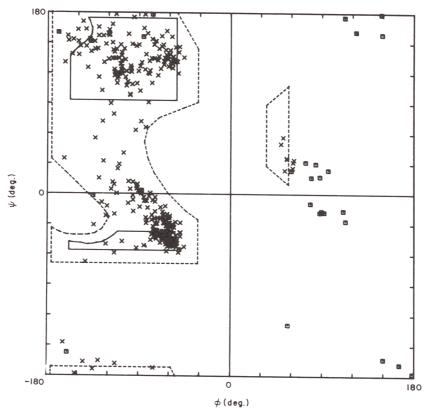


Figure 5. The φ, ϕ plot of the complex of thermitase with eglin-c in crystal form II. The φ, ϕ -angles of non-glycine residues are indicated by a cross and the glycine residues by a square. The continuous lines define the areas of fully allowed conformations with $\tau(N, C^z, C) = 110^\circ$ for the non-glycine residues. The broken lines show the regions obtained by relaxing the van der Waals' contact constraints as well as by allowing τ to increase to 115° (Ramakrishnan & Ramachandran, 1965).

tions from ideality as calculated by the program TNT (Tronrud *et al.*, 1987) are included. These numbers compare well to what has been obtained

Table 3

Energies of the final refined model of thermitase-eglin-c complex, form II, and the root-mean-square deviations from ideality

Term	Energy (kJ/mol)
Covalent bond	$0.85 \times 10^3 \ (0.33)^{\dagger}$
Bond angle	$0.21 \times 10^4 \ (0.58)$
Improper dihedral	$0.51 \times 10^3 \ (0.44)$
Dihedral	$0.14 \times 10^4 \ (0.91)$
Lennard-Jones	-0.10×10^{5}
Electrostatic	-0.12×10^{5}
	Root-mean-square deviation from ideality‡
Covalent bond (Å)	0.019
Bond angle (°)	2:73

16.77

0.026

0.047

0.126

Torsion angle (°)

Planar groups (Å)

Bad contacts (Å)

Trigonal atom non-planarity (Å)

for other proteins using more classical least-squares refinement techniques (e.g. see Tronrud et al., 1986). An impression of the accuracy of the model can be obtained from a $\sigma_{\rm A}$ plot (Read, 1986), see Figure 6. The overall root-mean-square co-ordinate error for the thermitase–eglin-c complex is 0·13 Å, as estimated from the slope of the $\sigma_{\rm A}$ plot.

Individual atomic isotropic temperature factors were refined using Agarwal's FFTREF (Agarwal, 1978). An average temperature factor of 11.9 Å^2 was obtained for the whole complex, including 227 water molecules. The average temperature factor of the 2004 non-hydrogen atoms of thermitase is 10.0 Å^2 , whereas 12.9 Å^2 was obtained as the average temperature factor for eglin-c, containing 522 non-hydrogen atoms. The 227 water molecules have an average B-factor of 24.6 Å^2 . The average main-chain temperature factors are depicted in Figure 7. Very low B-factors in thermitase are observed around the catalytic triad, Asp38, His71 and Ser225. In general, the temperature factors are higher when residues are more exposed to the solvent, as expected. The stabilizing effects of calcium binding are not all that clear from the temperature factors shown in Figure 7. Some residues, like Asp5 and Asp47, have indeed a lower B-factor than their surrounding residues. However, not all ligands from the loop 82 to 89 have a low B-factor. The residues surrounding the calcium-

[†]The average energy, i.e. per bond, bond angle, improper dihedral or dihedral, is given in parentheses.

[‡] As calculated by the TNT program (Tronrud *et al.*, 1987). No additional refinement with the TNT program of the model, at point H in Fig. 1, was carried out.

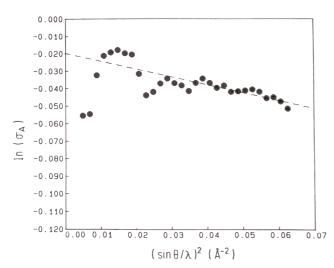


Figure 6. $\ln (\sigma_{\rm A})$ plotted against ($\sin \theta/\lambda$)². The broken line, fitted by a least-squares procedure, corresponds to a root-mean-square co-ordinate error (Read, 1986) of 0·13 Å. The 1st 2 points were not taken into account in the least-squares fitting.

binding sites will be discussed in more detail in section (d), below.

The lowest B-factors for eglin-c are observed in the inhibiting loop, residues 41I to 46I. A second region with low B-factors in eglin-c is the stretch 51I to 57I. This stretch contains the inhibiting loop stabilizing residues Arg51I and Arg53I. On average, eglin-c shows a slightly larger flexibility than thermitase, as reflected in the average temperature factors, $12 \cdot 9 \text{ Å}^2$ and $10 \cdot 0 \text{ Å}^2$, respectively. This difference in temperature factors between protease and inhibitor is not as large as in the case of subtilisin Carlsberg–eglin-c (Bode $et\ al.$, 1987), where values of $10 \cdot 0 \text{ Å}^2$ for subtilisin Carlsberg and $16 \cdot 5 \text{ Å}^2$ for eglin-c were reported. Although the temperature factors have been refined without restraints, they can clearly be correlated with specific aspects of the atomic model of the complex.

There are two *cis* peptide bonds in the structure of thermitase; both involve prolyl residues. In one case, the peptide precedes a prolyl residue, which is rather common. In the other case, the peptide follows a prolyl residue, which is very rarely observed. Figure 8 shows the two cis peptide bonds. The cis peptide bond between Pro214 and Thr215 has been observed for subtilisin Carlsberg between equivalent residues Pro210 and (McPhalen et al., 1986). In the two proteins, the cis peptide bond facilitates a reverse turn at the surface of the protein. The other cis peptide bond found in thermitase is in common with subtilisin Carlsberg and subtilisin BPN' (Bode et al., 1987; McPhalen & James, 1988) and precedes a prolyl residue. This cis proline, Tyr171-Pro172, is shown in Figure 8(b) with its electron density. This cis peptide bond can be seen to be required for a short reverse turn in the main chain. This time, the turn is relatively more buried inside the protein. For both cis peptide bonds, the electron density is unambiguous.

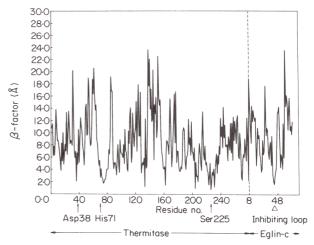


Figure 7. The main-chain temperature factors (A^2) , averaged over N, C^{α} , C and O, are given as a function of residue number for thermitase and eglin-c. The catalytic triad residues of thermitase are indicated in the plot, as well as the residues of eglin-c directly involved in binding to thermitase. The residue numbering of eglin-c starts with 8, because the 1st 7 residues are invisible in the map and are omitted from the model.

While the density is well-defined at most places, in a few instances this is not the case. This involves Asp60, Ser138, Val199 and Pro229. Asp60 will be discussed in section (d), below. Ser138 appears to be somewhat disordered in crystal form II, though in crystal form I it has well-defined electron density. Pro229 lacks electron density for its C⁷ atom. In crystal form I, the full electron density for a prolyl residue is visible. Both crystal forms show electron density that could accommodate a tryptophanyl residue at the position of Val199. The published amino acid sequence might have to be corrected for residue 199 of thermitase.

There are several novel features in the model of thermitase in comparison to the other subtilisins. The largest differences in main-chain fold can be found at insertion and deletion sites. Two of these new features, the N and C-terminal extensions, are highlighted here, see Figure 9. The N-terminal extension, residues Tyrl to Tyr7, is clearly visible in the electron density map, as shown for residues 3 to 8 in Figure 9(a). Residues Tyrl to Gln12 are totally rearranged in thermitase with respect to the other subtilisins. A short helical turn consisting of only four residues is formed by Tyr7 to Ser10, following the definition of Kabsch & Sanders (1983). The N-terminal calcium ligand, Gln2 in subtilisin Carlsberg and BPN', is replaced by Asp5 from the N-terminal extension in thermitase. The terminal α-amino group of thermitase is nicely stabilized by a hydrogen bond to O^{δ} of Asp25 (d = 2.82 Å).

The C-terminal extension, Tyr279, is shown in Figure 9(b). Again, the electron density shows the additional residue unambiguously. The C-terminal Tyr279 is stabilized by a hydrogen bond between its hydroxyl oxygen and the side-chain nitrogen of Asn254.

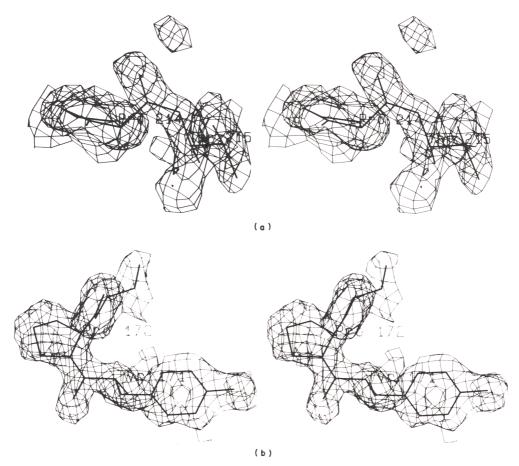


Figure 8. The cis peptide bonds in thermitase. (a) The cis Thr bond, Pro214–Thr215, is shown in its final $2mF_{\rm o}$ - $DF_{\rm c}$, $\alpha_{\rm c}$, electron density. (b) The cis Pro bond, Tyr171–Tyr172, and the electron density for the 2 residues is shown. The electron densities were contoured at 1·1 σ .

Though there are several novel features in the model, the thermitase core resembles subtilisin Carlsberg to a large extent. The root-mean-square difference in C^{α} co-ordinates between thermitase and subtilisin Carlsberg is 2·02 Å for 267 residues and 1·99 Å for 267 residues between thermitase and subtilisin BPN'. The root-mean-square differences drop to 0·31 Å and 0·50 Å if only the best 148 or 203 superimposable C^{α} atoms are taken into account for subtilisin Carlsberg and BPN', respectively. These numbers correspond well to what Chothia & Lesk (1986) predict from the degree of amino acid simi-

larity. The structure of the complex thermitase–eglin-c allows a detailed comparison with the other subtilisins for evaluation of the insertions, deletions and many substitutions.

(c) The binding of eglin-c

By far the largest difference between the thermitase–eglin-c complexes of TE1 and TE2 is the orientation of the inhibitor core with respect to the protease. Figure 10 shows the C^{α} tracings of TE1 and TE2 after superposition of the thermitase

Table 4

The main-chain torsion angles, φ and ψ , of the inhibiting loop of eglin-c in complex with thermitase, TE1 and TE2, and subtilisin Carlsberg, CE

	Residue	P42I	V43I	T44I	L45I	D46I	L47I	R48I
Complex	P-site	P4	Р3	P2	P1	P_1'	P_2'	P_3'
TE1 TE2 CE		$-68/152 \\ -71/136 \\ -72/138$	$-129/152 \\ -138/153 \\ -141/165$	$-65/133 \\ -64/143 \\ -58/142$	$-99/51 \\ -105/40 \\ -116/47$	-103/166 $-85/164$ $-98/169$	-108/129 $-112/110$ $-118/106$	-137/121 $-110/119$ $-117/109$

 $[\]omega_{\rm TE1} \; ({\rm L45I-D46I}) = 173; \; \omega_{\rm TE2} \; ({\rm L45I-D46I}) = 167; \; \omega_{\rm CE} \; ({\rm L45I-D46I}) = 161.$

The ω angles of the scissile peptide plane are also listed for the 3 complexes. Data concerning subtilisin Carlsberg–eglin-c are taken from Bode et al. (1986). All angles are in degrees.

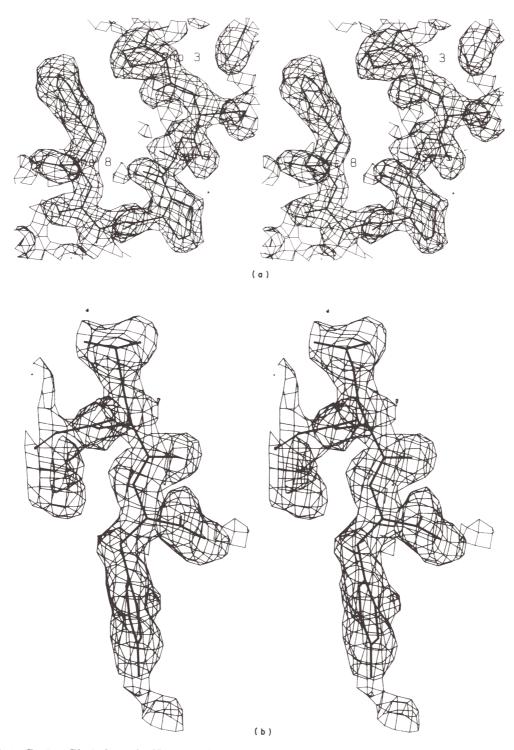


Figure 9. (a) Pro3 to Phe8, from the N-terminal extension of thermitase, shown in the $2mF_{\rm e}$ - $DF_{\rm e}$, $\alpha_{\rm e}$, electron density. (b) The C-terminal residues Gln278 and Tyr279 of thermitase shown with the electron density. The electron densities are contoured at 1·1 σ .

molecules only. The inhibiting loop of eglin-c, residue Ser41I to Leu47I, is in equivalent positions in both structures. The root-mean-square difference between equivalent C^α atoms in this loop is 0·19 Å. The cores of eglin-c, consisting of Lys8I to Gly40I and Arg48I to Gly70I, differ by a rigid body rotation of 10·4°. The root-mean-square difference between the 56 C^α atoms of the two cores is 0·42 Å

after applying the $10\cdot4^{\circ}$ rotation. This shows that the cores by themselves, as well as the inhibiting loops by themselves, are virtually identical and that basically the motion is that of a rigid body with hinges near Ser41I and Leu47I. The structure of eglin-c complexed with subtilisin Carlsberg (Bode *et al.*, 1986; McPhalen *et al.*, 1985a) shows, yet again, a slightly different orientation of the eglin-c core. It is

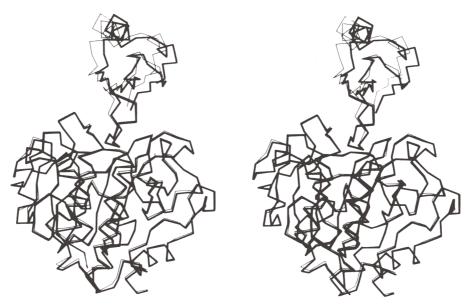


Figure 10. Comparison of the 2 thermitase–eglin-c complexes. The C^{α} tracing of the complex thermitase–eglin-c as observed in crystal form II (TE2) shown in thick lines. The complex in crystal form I (TE1) shown in thin lines. The structure of TE1 was determined by Gros *et al.* (1989) and refined to an *R*-factor of 17.9%. Using the TE2 model, the TE1 model could be improved only slightly. The resulting *R*-factor was 17.5%.

likely that eglin-c is, to some extent, flexible in solution, giving rise to different orientations of its core relative to the protease.

The main-chain torsion angles, φ and ϕ , of the inhibiting loop of eglin-c in TE1, TE2 and in the complex of eglin-c with subtilisin Carlsberg (Bode et al., 1986) are given in Table 4. (For convenience, the complex of subtilisin Carlsberg with eglin-c will sometimes be referred to as CE in this discussion). Clearly, the inhibiting loop exhibits equivalent conformations in the three complexes. The largest

distortion from planarity for the scissile bond, Leu45I–Asp46I, is found in the complex with subtilisin Carlsberg. In the complexes with thermitase, a smaller deviation from planarity is observed. The distance between the O⁷ of Ser225 and the carbonyl carbon of the scissile bond is 2.91 Å and 2.77 Å in TE1 and TE2, respectively, whereas Bode *et al.* (1986) observe a distance of 2.81 Å in the complex of subtilisin Carlsberg with eglin-c. This indicates that proteolysis is stopped at the same point along the reaction pathway in all three complexes.

Table 5

Hydrogen bonding between protease and inhibitor in the complexes of thermitase with eglin-c (TE1 and TE2), and in subtilisin Carlsberg with eglin-c (CE)

				Сс	omplex		
Elg	gin-e		TE1		TE2		CE
Residue	Atom	Atom	Residue	Atom	Residu	e Atom	Residue
Gly40I Pro42I	0	NE1	Trp112 (2·78; 43·6)	NE1 OG	Trp112 (3·11; 51·5 Ser106 (3·46; 9·4)		Tyr104 (2·57; 26·2)
Val43I	N O	N O N	Gly110 (3·36; 52·3) Gly135 (3·22; 10·7) Gly135 (3·00; 31·4)	N O N	Gly110 (3·40; 31·8 Gly135 (2·88; 24·9	0	Gly102 (3·34; 35·5) Gly127 (2·97; 7·7)
Thr44I Leu45I	N N	0	Gly108 (3·02; 30·9) Ser133 (3·32; 16·4)	0	Gly135 (3·02; 21·4 Gly108 (2·90; 31·2 Ser133 (3·34; 23·3	0	Gly127 (2·94; 21·9) Gly100 (2·91; 24·7)
	0	ND2 N	Asn163 (2·76; 5·6) Ser225 (3·20; 16·6)	ND2 N	Asn163 (2·89; 11·2 Ser225 (3·00; 26·0		Ser221 (2·91; 54·8) Asn155 (2·72; 32·5) Ser221 (2·99; 20·7)
Leu47I Arg48I	N NH1	ОН	Ser222 (2·76; 29·4) Tyr213 (3·15; 58·5)	0	Ser222 (2·82; 31·7		Asn218 (2·90; 48·7)
Arg53I	NH2 NH2	0	Ser107 (3·47; 15·5)	0	Ser107 (3·36; 21·3)	O O	Asn62 (3·00; 19·5) Asn62 (3·31; 37·7)

The distance (Å) and angular deviation from 180° (°) between hydrogen donor and acceptor are given in parentheses. Only hydrogen bonds with a distance less than 3.5 Å and a deviation less than 60° from linearity are considered. These data were calculated using the program HBONDS within the molecular analysis and graphics program package WHATIF (G. Vriend, unpublished results).

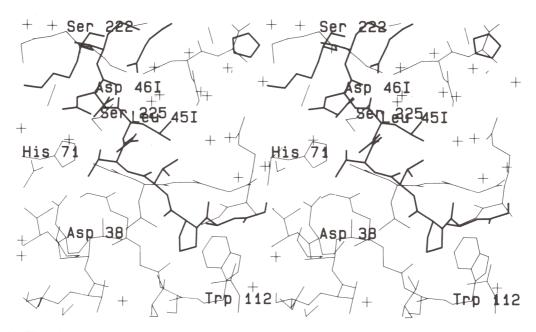


Figure 11. The inhibiting loop of eglin-c bound to thermitase in crystal form II. Residues of eglin-c are shown in thick lines. The active site region and the binding sites of thermitase are shown in thin lines. Labelled residues are discussed in the text. Water molecules are indicated by crosses.

Table 5 shows the hydrogen bonds identified in the current models between the inhibitor, eglin-c, and the two thermitases and subtilisin Carlsberg. The hydrogen-bonding network for Pro42I to Leu47I is very similar for the three cases. Also, the water-mediated contact between Thr44I, Asp46I and Arg51I of eglin-c and Asn62 in subtilisin Carlsberg (Bode et al., 1986) is observed in the two thermitase—eglin-c complexes. The following alterations in the hydrogen pattern can be noted (Table 5).

- (1) A slight difference in hydrogen bonding seems to occur for the main-chain nitrogen of Leu45I, when comparing subtilisin Carlsberg with the thermitase complexes. However, the co-ordinate differences between the three complexes are small for the nitrogen of Leu45I and its potential hydrogen acceptors.
- (2) Tyr104 in subtilisin Carlsberg is replaced in thermitase by Trp112. The hydrogen bond of the side-chain hydroxyl of Tyr104 with the carbonyl oxygen of Gly40I in the complex of subtilisin Carlsberg with eglin-c is neatly replaced by a hydrogen bond between the side-chain NH group of Trp112 with the same carbonyl oxygen of the inhibitor, in the complex of thermitase with eglin-c.
- (3) A rearrangement of hydrogen bonds occurs in the neighbourhood of Arg48I. The carbonyl oxygen of Asn69 in thermitase does not directly hydrogen bond to the inhibitor, unlike the equivalent Asn62 in subtilisin Carlsberg. Instead, it makes a relatively poor hydrogen bond with the hydroxyl of Tyr213. The hydroxyl of Tyr213 has, in turn, a possible hydrogen bond to the N⁷² of Arg48I.
- (4) In TE2, a water molecule mediates between

 N^{η_2} of Arg48I and $O^{\delta 1}$ of Asn69 and the mainchain nitrogen of His71. This water molecule is absent in both TE1 and CE. The insertion of a water molecule in TE2 at the interface of eglin-c with thermitase can be correlated with the rigid body displacement of eglin-c in TE2 compared to eglin-c in TE1 and CE.

The contacts of eglin-c with thermitase and subtilisin Carlsberg are given in Table 6. Of 29 residues of the proteases that are in close contact with eglin-c, 22 are identical. Six of the amino acid substitutions do not influence the P_3 to P_3' binding pocket directly. The seventh, Ser222 in thermitase (equivalent to Asn218 in subtilisin Carlsberg), is near the P_1' and P_2' binding pockets. Nonetheless, this amino acid difference between thermitase and subtilisin Carlsberg does not directly affect the binding, because the side-chains are pointing away from the binding site. Serine, instead of asparagine, at this position is important for the thermostability, as shown by Bryan et al. (1986).

There are significant differences in binding of eglin-c in the three complexes. This includes the Tyr104 to Trp112 difference at the S_4 site between subtilisin Carlsberg and thermitase. In the S_4 site, many more contacts of Pro42I with thermitase are observed than with subtilisin Carlsberg. As shown in Figure 11, Trp112 in thermitase stacks nicely with Pro42I, giving rise to many hydrophobic contacts between protease and inhibitor at the S_4 site.

At the S'₃ site, several van der Waals' contacts appear to be lost in the thermitase–eglin-c complexes. As pointed out earlier, the hydrogen bonds are also different at this site, but no amino acid differences are directly involved. The carbonyl oxygen of Asn69 in thermitase is clearly further away from the inhibitor than the equivalent oxygen

Number of contacts between eglin-c and thermitase in crystal form I (TEI) and crystal form II (TE2), and between eglin-c and subtilisin Carlsberg (CE) Table 6

,							1.5	Γ4	Г3	Γ_2	Υ ₁	Γ_1	F_2	F_3	F.4					
H	Eglin-c	Tyr		Leu		Gly	Ser	Pro	Val	Thr	Leu	Asp	Leu	Arg	Tyr	Arg		Arg		
		35	36	37	38 39	40	41	42	43	44	45	46	47	48	49	50 51	52	53	Totals in	s in
Carlsberg																		TEI	31 TE2	2 CE
Thr Asn u:										1,1,1	,	6		2,0,8				- 21	0 1	- ∞
Leu Con										10,10,9 $1,1,1$	1,1,1	2,2,3		2,0,0						
		3,1,4							2,5,5	2,4,3								1,1,1 3,3,3		
101 Ser 102 Glv			ψ.	2,2,0				1,1,2	0,0,1											
						,		3,3,0										# 60		
						4,1,2		15,11,3 $2,2,2$										91		
										0,0,1	1,1,0							1 —		
								0,1,0	4,4,5		2,2,3							9		
						0,1,0	0,1,0	0,0,1	O, T, O		2,0,0							0		
						0,0,5												0		
											2,1,1							o 61		
											1,2,2 3,4,7	0.1.3						1		
											6,1,6		5,3,5					. rc		
204 Gly															0,6,0			0		
														3,3,0				ကင		
												2.2.3		0,±,1				o re		
											2,1,2		1,0,0					9		
)	1,2,1 9 12 10	6 6 6						_ =		
										•		0,0,1						0		
Totals in						1														
TE2		ئ 		N 61		- 4	0 -	2 29 26	9 01	14 16	25 29	9 1-	ග ග	c 1-	0 9			4 4		
		4		0		10	0	14	Ξ	15	27	12	9	13						

Two atoms are considered to be in contact when the distance between them is less than the sum of their van der Waals' radii plus 0.5 Å. If contacts occur, each position of this "contact matrix" contains 3 numbers: the 1st being the number of contacts between an inhibitor and a protease residue in TEI, the 2nd being the number in TE2, and 3rd in CE.

of Asn62 in subtilisin Carlsberg. The C^{α} atoms of the asparaginyl residues of thermitase and subtilisin Carlsberg are 0.68 Å apart after superposition. The preceding glycyl residues superimpose even worse, their C^{α} atoms deviate 2.3 Å after superposition. These differences between the proteases are the consequence of the new calcium-binding site in thermitase around residues 57 to 66.

Another interesting amino acid difference, concerning substrate specificity, between thermitase and the other subtilisins is located at the bottom of the S₁ site. In subtilisins Carlsberg and BPN', Gly166 is found at this position. In the complex of eglin-c with subtilisin Carlsberg, the P₁ pocket is filled with the side-chain of Leu45I and two water molecules. In thermitase, Gly166 is replaced by Asn170. No water molecule can be found in the complexes of thermitase and eglin-c in the P₁ pocket, but also no direct interaction occurs between the side-chains of Leu45I and Asn170. It is likely that the preference for an arginyl residue in the S_1 site (Brömme & Kleine, 1984) may be explained by specific hydrogen bonds of the guanidyl group of arginine with the side-chain of asparagine at position 170 in thermitase.

(d) The three ion-binding sites

Biochemical data show that thermitase can bind three calcium ions (Frömmel & Höhne, 1981). All three ion-binding sites have been identified in the electron density map. The residues involved in chelating these ions are listed in Table 7. Two sites, Ca1 and Ca3, are in common with subtilisin Carlsberg and BPN'. In spite of large sequence differences, especially at the N terminus, the chelating ligands are conserved to a considerable extent in

these sites. Site Ca2 in thermitase is a newly observed calcium-binding site for the subtilisins. Table 8 gives the distances between ligand and cation for the three calcium-binding sites of both crystal forms of thermitase–eglin-c. Figure 12 shows the three ion-binding sites and their surroundings in crystal form II of the thermitase–eglin-c complex.

The strong calcium-binding site of thermitase, Ca1, has an octahedral arrangement of ligands (Fig. 12(a)). The ligands are provided by three different parts of the protein (Table 7): (1) Asp5 from the N-terminal extension; (2) Asp47; and (3) residues 82, 85, 87 and 89 from the "Ca1-embracing loop". In addition to links via the calcium ion, the N-terminal extension and the Ca-1 embracing loop are held together by hydrogen bonds between Asn4 and Asp5 of the extension, and Asn84 of the embracing loop. These interactions between different parts of the protein may contribute to its thermostability.

When comparing crystal forms I and II, no large differences are apparent in the Ca1 site (Table 8). However, Thr87 appears to be 0·18 Å further away from the calcium ion in TE2 than in TE1. This does not correspond with the calcium concentrations in the crystallization medium, which is higher in TE2. The crystallographic B-factors of the ions are in both forms equal to the lower limit, which was set to 0·5 Ų in the refinement. This indicates that the Ca1 site is most likely fully occupied in both crystal forms.

Comparison of the Cal site in thermitase with the equivalent, "strong", calcium-binding site in subtilisin Carlsberg and BPN' (Table 7) shows two differences. First, the Cal-embracing loop is one residue longer in thermitase. Asn84 in thermitase loops out in comparison to subtilisin Carlsberg, thus

Table 7
Alignment of the thermitase sequence with subtilisin Carlsberg in the regions involved in chelating calcium

A. Calcium	n-bina	ling s	ite 1																			
	1	2°	3	4	5	6	7	8	9	10	11	12	 47	 82	83	84	85	86	87	88	89	
Therm.	Y	\mathbf{T}	P	N	D	P	Y	\mathbf{F}	\mathbf{S}	\mathbf{S}	\mathbf{R}	Q	D	V	\mathbf{T}	N	N	\mathbf{S}	\mathbf{T}	G	Ι	
Carlsberg				A	Q					\mathbf{T}	V	P	D	L		D	N	\mathbf{T}	Τ	G	V	
				1	2					3	4	5	 41	 74		75	76	77	78	79	80	
					*								*	*			*		*		*	
					\mathbf{S}								\mathbf{S}	M			\mathbf{S}		\mathbf{M}		\mathbf{M}	
B. Calciun	n-bina	ling s	ite 2†																			
		57	58	59	60	61	62	63	64	65	66											
Therm.		D	\mathbf{F}	V	D	N	D	\mathbf{S}	\mathbf{T}	P	Q											
Carlsberg		\mathbf{S}	\mathbf{F}	V	A	G	\mathbf{E}	A	Y	N	T											
		49	50	51	52	53	54	55	56	57	58											
		*			*		*		*		*											
		\mathbf{S}			\mathbf{S}		\mathbf{S}		M		\mathbf{S}											
C. Calcium	n-bind	ing si	ite 3																			
		173	174	175	176	177	178		201													
Therm.		A	Y	Y	\mathbf{S}	N	A		\mathbf{D}													
Carlsberg		A	K	Y	D	\mathbf{S}	V		\mathbf{E}													
		168	169	170	171	172	173		196													
		*		*	*		*		*													
		M		M	M		M		\mathbf{S}													

Residues involved in calcium binding are indicated *. S below the asterisks indicates that a side-chain oxygen is involved in ligating to Ca²⁺, and M indicates that the main-chain carbonyl acts as a ligand.

[†] Here, S and M refer to thermitase only. This calcium-binding site is absent in subtilisin Carlsberg.

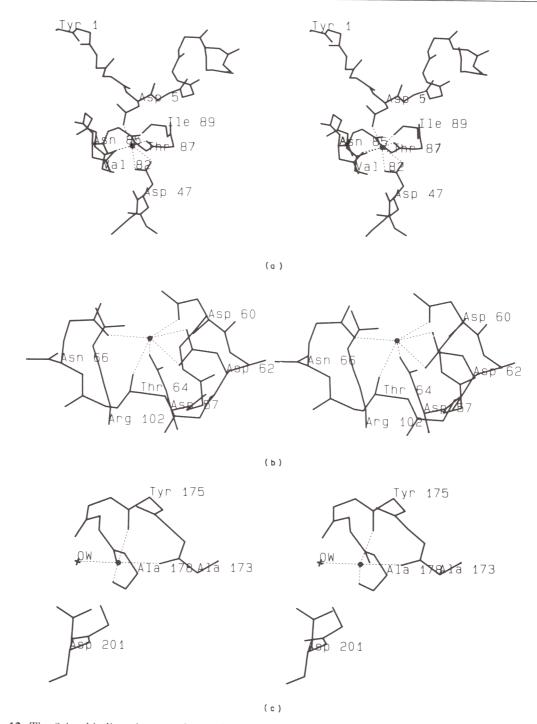


Figure 12. The 3 ion-binding sites, as observed in TE2: (a) the strong calcium-binding site (Ca1); (b) the medium-strength calcium-binding site (Ca2); and (c) the weak calcium-binding site (Ca3). The ion positions are indicated by a dot. The weak calcium-binding site is thought not to be occupied by Ca^{2+} but by K^+ or Na^+ . The cross indicates the position of a solvent molecule.

providing a number of possible hydrogen bonds with the N-terminal extension. Second, the chelating side-chain carbonyl oxygen of Gln2 in subtilisin Carlsberg is replaced by the carboxylate oxygen of Asp5 in thermitase. This suggests that this site binds calcium with greater affinity in thermitase than in subtilisin Carlsberg.

The novel calcium-binding site Ca2 could be identified on the basis of (1) a low *B*-factor, (2) short

distances to a number of ligands and (3) a number of hydrogen bond acceptors as ligands (see Table 8 and Fig. 14). This calcium-binding site (Fig. 12(b)) shows, in first approximation, an octahedral ligand environment. One ligand position is possibly occupied by a very weakly bound water molecule, which has not been taken into account in the refinement. This water molecule is not displayed in Figure 12(b). The protein ligands are: four carboxyl

oxygen atoms provided by Asp57, Asp60 and Asp62; the side-chain carbonyl oxygen of Gln66; and the main-chain carbonyl oxygen of Thr64. This calcium-binding site is located about 13 Å away from the active site and is positioned close to the $\rm P_2$ pocket. It may therefore exert some effect on the binding of substrate and inhibitor.

At this Ca2 site, differences between the two thermitase-eglin-c crystal forms might be significant. First, the temperature factors of the calcium ion at site Ca2 show a large difference between TE1 and TE2, 11·02 Å² and 0·5 Å², respectively. This suggests that more Ca²⁺ is bound in TE2, as expected from the crystallization conditions. Second, in crystal form II, Asp57 and Asp62 are closer to the calcium ion in comparison to crystal form I (Table 8). Third, Asp60 and Arg102 are 0.85 Å further apart in TE2. Fourth, in the electron density of crystal form II, a new feature becomes visible. The mid-point of this piece of electron density is 2.44 Å away from the calcium ion, a relatively good Ca²⁺ ligand distance. The distance to $O^{\delta 1}$ and $O^{\delta 2}$ of Asp60 is, however, only 2.17 Å and 2.23 Å. This would be too close for a water molecule. However, inspecting the potential hydrogen donors and acceptors, this position could well be occupied by a water molecule. A different explanation involves a second conformation of Asp60. One of the O^{δ} atoms could possibly occupy the additional piece of electron density. This would require some atoms to shift by at least 2 Å. This explanation agrees with the fact that weak electron density is observed at the main chain of Asp60 (see Fig. 14). Moreover, it does not involve too close contacts between atoms. Therefore, we have possibly observed two significantly distinct conformations of Asp60 in this calcium binding site of thermitase in crystal form II.

Only in thermitase, residues 57 to 66 form a calcium-binding site (Ca2). The other subtilisins, subtilisin Carlsberg, BPN' and proteinase K, show a distinctly different conformation of this loop. Also, a very low level of sequence homology is observed between thermitase and the other subtilisins in this loop, see Table 7 (Meloun et al., 1985). It is interesting that the only single residue insertion of subtilisin BPN' versus subtilisin Carlsberg occurs in this loop. Furthermore, proteinase K exhibits no sequence identity at all with either subtilisin Carlsberg, BPN' or thermitase in this region. This shows that the evolution of this loop strongly diverged for the three types of proteins with the thermostable variant, thermitase, having a new calcium-binding site (Ca2) introduced in the structure at this site.

Binding site Ca3 is displayed in Figure 12(c) and (d), the distances of its geometry are given in Table 8. This site resembles the weak calciumbinding site of subtilisin Carlsberg and BPN'. There are two possible cation positions in this site, as shown by Pantoliano *et al.* (1988). The position near the carbonyl groups of Ala173, Tyr175 and Ala178

Table 8
Interatomic distances in the 3 ion-binding sites in the 2 crystal forms of thermitase-eglin-c, TE1 and TE2, with their differences

A. Geometry	y of site Ca1: see a	lso Fig. 13(a)			
Label	Atom 1	Atom 2	d_{TE1} (Å)	d_{TE2} (Å)	Δ (Å)
a	Ca ²⁺	O^{δ} , Asp5	2.37	2.38	+0.01
b	Ca ²⁺	O, Ile89	2.42	2.43	+0.01
e	Ca ²⁺	O, Thr87	2.36	2.54	+0.18
d	Ca ²⁺	$O^{\delta 1}/O^{\delta 2}$, Asp47	2.48/2.60	$2 \cdot 42 / 2 \cdot 56$	-0.04/-0.06
e	Ca ²⁺	O^{δ} , Gln85	2.47	$2.57^{'}$	$+0.10^{'}$
f	Ca ²⁺	O, Val82	2.19	2.30	+0.11
$B_{\text{Ca}^2+, \text{TE}1} =$	0.5 Å ² ; B _{Ca²⁺, TE2}	$=0.5 \text{ Å}^2.$			
B. Geometry	y of site Ca2: see a	lso Fig. 13(b)			
Label	Atom 1	Atom 2	d_{TE1} (Å)	$d_{\rm TE2}$ (Å)	Δ (Å)
a	Ca^{2+}	O^{δ} , Asp60	2.84	2.90	+0.06
b	Ca ²⁺	O^{δ} , Asp57	2.68	2.41	-0.27
e	Ca ²⁺	$O^{\delta 1}/O^{\delta 2}$, Asp62	2.56/2.56	2.33/2.44	-0.23/-0.12
d	Ca^{2+}	O, Thr64	$2.50^{'}$	2.46	-0.04
e	Ca ²⁺	O^{δ} , Gln66	2.56	2.52	+0.04
$B_{\text{Ca}^{2+}, \text{ TE1}} =$	= 11·02 Å ² ; B _{Ca²⁺, 1}	$_{\text{TE2}} = 0.5 \text{ Å}^2.$			
C. Geometry	y of site Ca3: see a	lso Fig. $13(c)$ and (a)	<i>!</i>)		
Label	Atom 1	Atom 2	d_{TE1} (Å)	d_{TE2} (Å)	Δ (Å)
a	Na^+	O, Tyr175	2.73	2.77	+0.04
b	Na^+	O, Ala178	2.48	2.64	+0.16
c	$\mathrm{Na^+}$	O, Ala173	2.67	2.58	-0.09
d	Na^+	O^{δ} , Asp201	3.24	3.35	+0.11
e	Na^+	OW	3.37	2.48	-0.89
f	ow	$O^{\delta 1}/O^{\delta 2}$, Asp20.	1 2.96/3.42	3.21/3.48	+0.25/+0.06
g	ow	O, Ser176		3.17	
g h	OW	N^{η} , Arg248		3.01	
B _{cation, TE1} =	= 9.96 Å ² ; B _{cation, T}	$T_{E2} = 15.03 \text{ Å}^2.$			

The labels correspond to the labels given in Fig. 13.

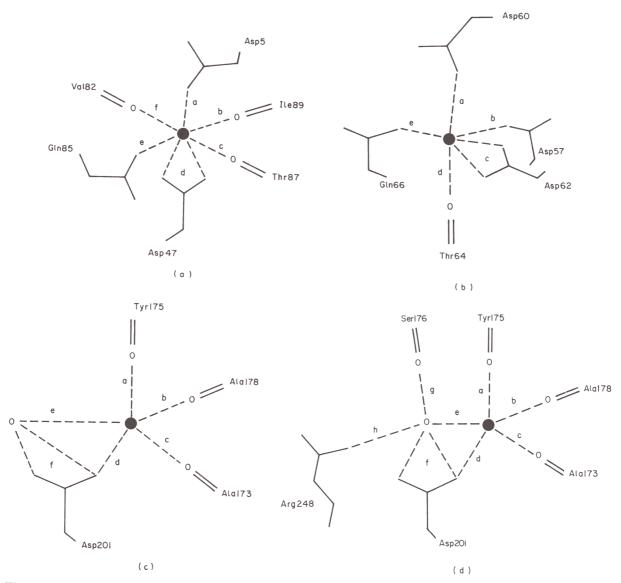


Figure 13. A representation of the ion-binding sites in both crystal forms (TE1 and TE2) of thermitase. Ion-to-ligand interactions are indicated with broken lines and with labels that refer to Table 8. a, The strong (Ca1) calcium-binding site as it is present both in TE1 and TE2; b, the medium-strength (Ca2) calcium-binding site, as seen in both TE1 and TE2; c, the weak (Ca3) calcium-binding site as observed in TE1; d, the weak (Ca3) calcium-binding site in TE2.

seems to be occupied by an ion in both TE1 and TE2 (see Table 8). However, there are indications that a calcium ion is not bound here, since the ligand-to-ion distances correspond better to K^+ or Na^+ binding. At the second position, near Asp201, a water molecule is bound to the protein, as can be deduced from the distances in Table 8. The position of this water molecule is significantly different in both thermitase structures, see also Figure 13(c) and (d). As a consequence, the local hydration shells are totally differently structured.

4. Discussion

We have presented the crystallographic refinement of the thermitase–eglin-c complex in crystal form II. The strategy employed here in the

molecular dynamics crystallographic refinement is able to perform large corrections, as shown previously by Fujinaga et al. (1989) and Gros et al. (1989). This compares well with the results obtained when using the simulated annealing protocol (Brünger, 1988). The rigid body displacement of the inhibitor core by $10\cdot4^{\circ}$ was corrected fully automatically. As shown, some entire residues were shifted by about 3 Å. No strict definition of rigid bodies or hinge bending regions is necessary when a crystallographic molecular dynamics procedure is used, as compared to the least-squares rigid body refinement programs like CORELS (Sussmann et al., 1977). Therefore, one may even be unaware of a rigid body displacement and be able to refine the structure properly.

Using the molecular dynamics crystallographic

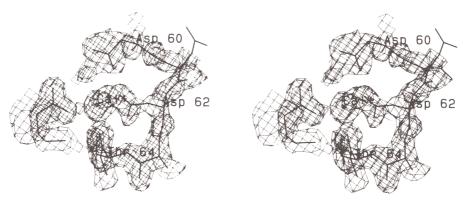


Figure 14. The $2mF_o$ - DF_c , α_c , electron density map around the calcium ion in Ca2, as observed in crystal form II of the thermitase–eglin-c complex. The position of the calcium ion is indicated by a cross. Additional electron density is visible near $O^{\delta 1}$ and $O^{\delta 2}$ of Asp60. This might be the result of an additional water molecule or of a different orientation of the carboxylate group.

refinement procedure, an R-factor of $16\cdot5\%$, for data between $8\cdot0$ and $1\cdot98$ Å and $|F_{\rm obs}| > \sigma_{\rm |obs|}$, was obtained readily. Only two minor model building sessions, which included water positioning, were required in the process of refinement. The inserted water molecules behaved properly during the molecular dynamics refinement at 300 K, without any need for positional restraints. The stereochemical indicators, as root-mean-square deviations from ideality and the φ, ϕ plot, indicated no serious model errors. Furthermore, a low, 0·13 Å, estimated root-mean-square co-ordinate error was obtained.

The thermitase model, obtained from crystal form II, resembles the model from crystal form I to a large extent. Both models show the novel N-terminal arrangement. However, this arrangement is absent in the structure presented by Teplyakov et al. (1989). The N terminus, the α -NH₃⁺ of Tyrl, is nicely hydrogen-bonded to Asp25. As discussed elsewhere (Gros et al., 1989), the loop 24 to 26 in thermitase is significantly differently arranged than the equivalent loops in subtilisins Carlsberg and BPN', which are one residue longer. This facilitates hydrogen bonding to the extended N terminus. The N-terminal segment is further stabilized by calcium binding to the strong calcium-binding site Ca1 by O^{δ} of Asp5. Furthermore, Asn84from Ca²⁺-embracing loop of Cal makes a number of hydrogen bonds with Asn4.

A detailed comparison of the structures of subtilisin Carlsberg, subtilisin BPN' and proteinase K with the structure of thermitase should reveal the importance of the numerous differences between these structures. Amino acid substitutions, like Asn to Ser at position 222 in thermitase, position 218 in subtilisin BPN', most likely increase the intrinsic stability of the protein (Bryan et al., 1986). Other differences, e.g. the four amino acid deletion between Pro169 and Asn170 in thermitase, may reduce autoproteolysis. The comparison of the subtilisins, including the thermostable variant thermitase, with respect to thermostability and autodegradation will be discussed elsewhere.

The different orientations found for the core of

eglin-c in the two complexes with thermitase and in the complex with subtilisin Carlsberg clearly reflects the flexibility of the inhibitor in solution. Comparing the two thermitase-eglin-c complexes, the effects of the different orientation of eglin-c can be studied. The P_4 to P'_2 residues of the inhibiting loop seem to be unaffected by the rigid body motion of the core. At the S'₃ site, on the other hand, an additional water molecule is found at the protease-inhibitor interface in the complex of crystal form II. This indicates that the interactions between thermitase and eglin-c are weak at the S'₃ site, and possibly there exists an equilibrium between the two forms in solution. The weak interaction found at the S'₃ site corresponds with the biochemical data on thermitase presented by Brömme et al. (1986), indicating that elongation of the substrate peptide chain up to the third amino acid residue on the C-terminal side does not significantly enhance the hydrolysis rate.

The differences in the substrate-binding pockets between thermitase and subtilisin Carlsberg explain some of the differences in substrate binding specificity. The comparison of the binding of the inhibitor eglin-c to thermitase and subtilisin Carlsberg shows that significant differences occur at the S₄ site. The preference of thermitase for aromatic residues at the P₄ site of substrates (Brömme & Kleine, 1984) can be explained by stacking of the aromatic residue of the substrate with Trp112 of thermitase. Additional differences are observed at the S_3' site. However, these are more difficult to evaluate, because they are believed to be relatively weak, as discussed above. The interesting difference in amino acid composition of the S₁ site between subtilisin Carlsberg and thermitase, Gly166 and Gln170, respectively, does not change the binding of eglin-c. However, an arginyl residue at P_1 could hydrogen bond with the side-chain of Gln170 of thermitase; this interaction is absent when bound to subtilisin Carlsberg.

Analysing the ion-to-ligand distances at the three ion-binding sites in thermitase shows that two of the three calcium-binding sites are occupied by a

calcium ion in both crystal forms. The weak calcium-binding site (Ca3) is most likely occupied by a monovalent ion. In analogy with the results of Pantoliano et al. (1988), we expect that, when the calcium concentration is increased, a calcium ion will bind to Asp201 at this site. Two stages in the process of binding Ca²⁺ have been observed at the medium-strength calcium-binding site Binding of Ca²⁺ possibly induces a localized conformational change at this site. A concerted set of small differences over a larger area, as described by Bajorath et al. (1989), is not observed between the current models of thermitase. The binding of calcium at the strong calcium-binding site (Ca1) seems to be improved in thermitase in comparison to subtilisins Carlsberg and BPN'. An additional, negatively charged ligand is introduced by replacing Glu2 in subtilisin Carlsberg and BPN' by Asp5 in thermitase. This may be an important feature of thermostability of thermitase induced by calcium in the presence of chelating agents, such as EDTA. In the 0 to 5 $\rm mm\text{-}CaCl_2$ range, thermitase is stabilized by binding of Ca2+ at the novel, medium-strength calcium-binding site.

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References

- Agarwal, R. C. (1978). Acta Crystallogr. sect. A, 34, 791–809.
- Bajorath, J., Hinrichs, W. & Saenger, W. (1988). Eur. J. Biochem. 176, 441–447.
- Bajorath, J., Raghunathan, S., Hinrichs, W. & Saenger, W. (1989). Nature (London), 337, 481–484.
- Berendsen, H. J. C., Postma, J. P. M., Gunsteren, W. F. van, Dinola, A. & Haak, J. R. (1984). J. Chem. Phys. 81, 3684–3690.
- Bernstein, F. C., Koetzle, T. F., Williams, G. J. B., Meyer, E. F., Jr, Brice, M. D., Rodgers, J. R., Kennard, O., Shimanouchi, T. & Tasumi, M. (1977). J. Mol. Biol. 112, 535–542.
- Betzel, C., Pal, G. P. & Saenger, W. (1988a). *Acta Crystallogr. sect. B*, **44**, 163–172.
- Betzel, C., Pal, G. P. & Saenger, W. (1988b). Eur. J. Biochem. 178, 155–171.
- Bode, W., Papamokos, E., Musil, D., Seemüller, U. & Fritz, H. (1986). *EMBO J.* **5**, 813–818.
- Bode, W., Papamokos, E. & Musil, D. (1987). Eur. J. Biochem. 166, 673–692.
- Bott, R., Ultsch, M., Kossiakoff, A., Grayear, T., Katz, B. & Power, S. (1988). J. Biol. Chem. 263, 7895–7906.

- Brömme, D. & Kleine, R. (1984). Curr. Microbiol. 11, 93–100.
- Brömme, D., Peters, K., Fink, S. & Fittkau, S. (1986). *Arch. Biochem. Biophys.* **244**, 439–446.
- Brünger, A. T. (1988). J. Mol. Biol. 203, 803-816.
- Brünger, A. T., Kuriyan, J. & Karplus, M. (1987). Science, 235, 458–460.
- Bryan, P. N., Rollence, M. L., Pontoliano, M. W., Wood, J., Finzel, B. C., Gilliland, G. L., Howard, A. J. & Poulos, T. L. (1986). *Proteins*, 1, 326–334.
- Chothia, C. & Lesk, A. M. (1986). EMBO J. 5, 823–826.
 Dauter, Z., Betzel, C., Höhne, W. E., Ingelman, M. & Wilson, K. S. (1988). FEBS Letters, 236, 171–178.
- Drenth, J., Hol, W. G. J., Jansonius, J. N. & Koekoek, R. (1972). *Eur. J. Biochem.* **26**, 177–181.
- Finzel, B. C. (1987). J. Appl. Crystallogr. 20, 53-55.
- French, S. & Wilson, K. S. (1978). Acta Crystallogr. sect. A, 34, 517–525.
- Frömmel, C. & Höhne, W. E. (1981). *Biochim. Biophys. Acta*, **670**, 25–31.
- Frömmel, C., Hausdorf, G., Höhne, W. E., Behnke, U. & Ruttlof, H. (1978). Acta Biol. Med. Germ. 37, 1193–1204.
- Fujinaga, M., Gros, P. & van Gunsteren, W. F. (1989).
 J. Appl. Crystallogr. 22, 1–8.
- Gros, P., Fujinaga, M., Dijkstra, B. W., Kalk, K. H. & Hol, W. G. J. (1989). *Acta Crystallogr*. In the press.
- Hall, S. R. & Stewart, J. M. (1987). Editors of XTAL2.2 User's Manual, Universities of Western Australia and Maryland.
- Hendrickson, W. A. & Konnert, J. H. (1981). In Biomolecular Structure, Function, Conformation and Evolution (Srinivasan, R., ed.), Vol. 1, pp. 43–57, Pergamon Press, Oxford.
- Hermans, J. (1985). Molecular Dynamics and Protein Structure, Polycrystal Books Service, Western Springs, II.
- Jack, A. & Levitt, M. (1978). Acta Crystallogr. sect. A, 34, 931–935.
- Jones, T. A. (1978). J. Appl. Crystallogr. 11, 268–272.
- Kabsch, W. & Sanders, C. (1983). *Biopolymers*, **22**, 2577–2637.
- Karplus, M. & McCammon, J. A. (1981). CRC Crit. Rev. Biochem. 9, 293–349.
- Katz, B. A. & Kossiakoff, A. (1986). J. Biol. Chem. 261, 15480-15485.
- Kelders, H., Kalk, K. H. & Hol. W. G. J. (1989). J. Mol. Biol. 205, 615–616.
- Kleine, R. (1982). Acta Biol. Med. Germ. 41, 89-102.
- Laskowski, M., Jr & Kato, I. (1980). Annu. Rev. Biochem. 49, 593-626.
- McPhalen, C. A. & James, M. N. G. (1988). Biochemistry, 27, 6582–6598.
- McPhalen, C. A., Svendsen, I., Jonassen, I. & James, M. N. G. (1985a). Proc. Nat. Acad. Sci., U.S.A. 82, 7242–7246.
- McPhalen, C. A., Schnebli, H. P. & James, M. N. G. (1985b). FEBS Letters, 188, 55–58.
- McPhalen (1986). PhD thesis, University of Alberta, Canada.
- Meloun, B., Baudys, M., Kostka, K., Hausdorf, G., Frömmel, C. & Höhne, W. E. (1985). FEBS Letters, 183, 195–199.
- Neidhart, D. J. & Petsko, G. A. (1988). Protein Eng. 2, 271–276.
- Pantoliano, M. W., Whitlow, M., Wood, J. F., Rollence, M. L., Finzel, B. C., Gilliland, G. L., Poulos, T. L. & Bryan, P. N. (1988). Biochemistry, 27, 8311–8317.

- Ramakrishnan, C. & Ramachandran, G. N. (1965). Biophys. J. 5, 909-933.
- Read, R. J. (1986). Acta Crystallogr. sect. A, 42, 140-149.
- Seemüller, U., Fritz, H. & Eulitz, M. (1981). Methods Enzymol. 80, 804-816.
- Seemüller, U., Dodt, J., Fink, E. & Fritz, M. (1986). In Proteinase Inhibitors (Barrett, A. J. & Salvesen, G., eds), pp. 337–359, Elsevier, Amsterdam.
- Sussmann, J. L., Holbrook, S. R., Church, G. M. & Kim, S. H. (1977). Acta Crystallogr. sect. A, 33, 800-804.
- Teplyakov, A. V., Strokopytov, B. V., Kuranova, I. P., Popov, A. N., Harutyunyan, E. G., Vainshtein, B. K., Frömmel, C. & Höhne, W. E. (1986). Sov. Phys. $Crystallogr.~{\bf 31},~553-556.$
- Teplyakov, A. V., Kuranova, I. P., Harutyunyan, E. H., Frömmel, C. & Höhne, W. E. (1989). FEBS Letters, **244**, 208–212.
- Tronrud, D. E., Monzingo, A. F. & Matthews, B. W. (1986). Eur. J. Biochem. 157, 261–268. Tronrud, D. E., Ten Eyck, L. & Matthews, B. W. (1987).
- Acta Crystallogr. sect. A, 43, 489-501.
- van Gunsteren, W. F. & Berendsen, H. J. C. (1987). BIOMOS, Biomolecular Software, Laboratory of Physical Chemistry, University of Groningen, Groningen, The Netherlands.
- Voordouw, G., Milo, C. & Roche, R. S. (1976). Biochemistry, 15, 3716-3724.
- Wright, C., Alden, R. A. & Kraut, J. (1969). Nature (London), 221, 235–242.

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